Battery Safety

(and a few tips on using batteries to power your equipment)

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Agenda

I will focus on safety (and a few operational) issues that arise with three different battery chemistries:

- Lead-acid
- Lithium
- Alkaline

However, a lot of what I have to say about lead-acid batteries also applies to other types, so when we get to lithium and alkaline batteries, I will concentrate on the differences.

I'm not going to talk about nickel-based batteries today, since those are almost always fully integrated with the equipment.

Speaker Notes

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Three safety issues with deep-cycle lead-acid batteries

- Premium deep-cycle batteries are capable of delivering thousands of amperes into a short circuit when fully charged
- If the short-circuit is at a distance from the battery, intervening wires can be heated to incandescence, possibly starting a fire, melting insulation from other wires in the bundle
- Principal ingredients, lead and acid, both fall into the category of Very Nasty Stuff. It is important that these remain inside the battery and not be splattered around nearby people, equipment or the environment

- Can be connected in series or parallel to increase terminal voltage and/or capacity
 - ---but---
- Parallel connection should only be done with two batteries of the same design, age, and condition. Large currents can flow from one battery to the other if charge level and characteristics are not matched.
- Sooner or later, one of those batteries will fail. And the usual failure mode is a shorted cell. At this point, you will have a 12-V battery in parallel with a 10-V battery.
- So, fuse each battery separately, or buy a bigger battery!

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Two situations commonly occur that result in overcharging.

 First, the float voltage of a lead-acid battery is temperature-dependent. The voltage needs to be decreased by 3 mV/°C per cell above 25 °C and reduced when the temperature is below 25 °C.

Note: Manufacturers have different specifications. Concorde recommends compensation of <u>4 mV/°C</u>.

 3 or 4 mV doesn't sound like a lot, but a 12V leadacid battery has six cells, so we are really looking at a change of 18 to 24 mV/°C.



Example

12V SunXtender AGM battery (made by Concorde) with rated charging (float) voltage of 13.3 V (specified at 77 °F)

Temp (°F)	Temp (°C)	ΔT (°C)	ΔV (mV)	V _{Float} (V)	Result (when charged at 13.5 V)
77	25			13.3	battery fully charged as specified in data sheet
95	35	10	-240	13.06	overcharged – outgassing
32	0	-25	600	13.9	undercharged

The second issue is the shorted cell failure at the battery's end of life. Now the charger is applying 13.5 V to a 10-V battery instead of a 12-V battery. That is guaranteed to cause outgassing even in an air-conditioned room.

So, you need a battery charger that has two specific features:

- Temperature-compensated charging voltage
- Shorted-cell detection and shut-off



Two chargers I recommend are:

- BatteryMinder Model 12248 for 12V lead-acid batteries from 7 to 30 Ah
 - ---and---
- ProMariner Model 63120, also known as the ProNautic 1220P for powering a 100W HF station from a 80- to 100-Ah lead-acid battery.

Bothe models have the recommended safety features, and can be adjusted to a range of charging voltages for AGM, gel cell, and flooded lead-acid batteries.

If the battery temperature sensor is an extra-charge option, buy it!



Improvised power cord for ProNautic charger





Starting vs. deep-cycle batteries

- Starting batteries
 - Very high current to start engine, but very shallow discharge in normal use
 - Alternator maintains battery fully charged after short recovery period
 - Plates stacked to increase surface area. This lowers internal resistance and enables high current to be delivered
 - Weight is a key factor for vehicular batteries, so plates are thinner than deep-cycle batteries



Source: batteryuniversity.com

Starting vs. deep-cycle batteries

- Deep-cycle batteries
 - Peak current drain much lower in most applications
 - Battery is often the primary source of power to equipment, so battery may routinely experience deep discharge between charges
 - Thicker plates to maintain their geometry during deep discharge



Starting vs. deep-cycle batteries

- Plate thickness is the primary difference between starting and deep-cycle batteries
- Two plates, separated by a porous glass mat that is suffused with electrolyte, H₂SO₄
- During discharge, lead dioxide in the positive plate reacts with the sulfuric acid to produce lead sulfate.

$$PbO_{2} + HSO_{4}^{-} + 3H^{+} + 2e^{-} \rightarrow PbSO_{4} + 2H_{2}O$$

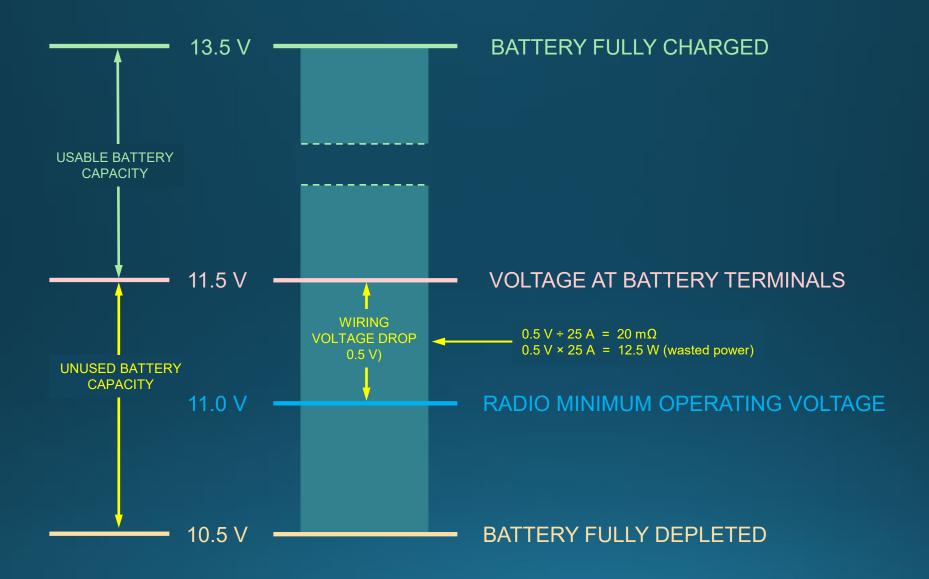
 At the same time, pure lead in the negative plate reacts with sulfate ions in the electrolyte to produce lead sulfate.

Voltage drop

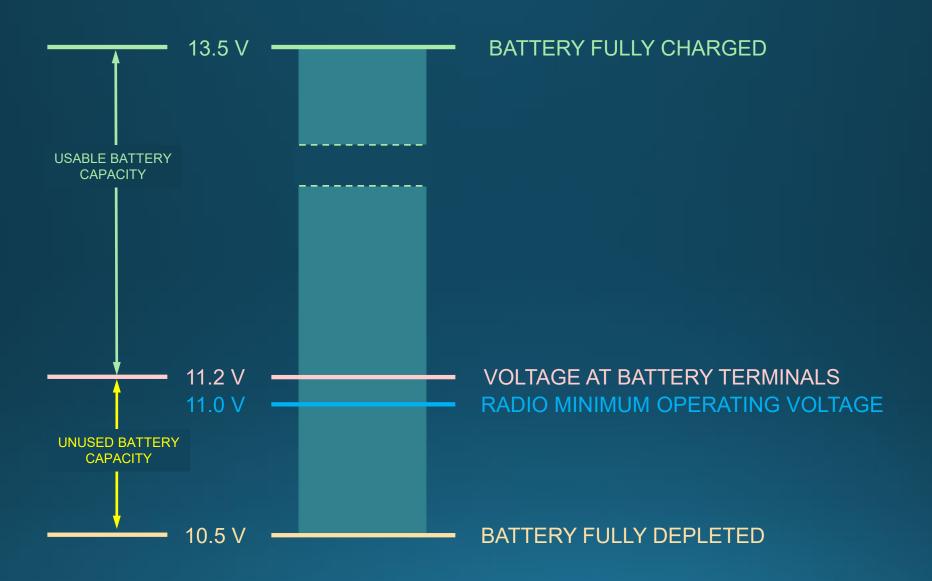
- Use heavy gauge wire from the battery to the radio, and keep the wires as short as possible to minimize voltage drop.
- In most radios, the output power drops with the supply voltage. So you may not get full power when the battery is substantially discharged. In addition, most radios will stop operating well before the battery reaches its terminal voltage of 10.5 V (beyond which it might be damaged.)
- If you are losing a half of a volt at the radio due to resistive loss in the battery cable, not only are you losing output power, but you are also losing run time (because the radio's minimum operating voltage is reached while the battery still has plenty of charge remaining).

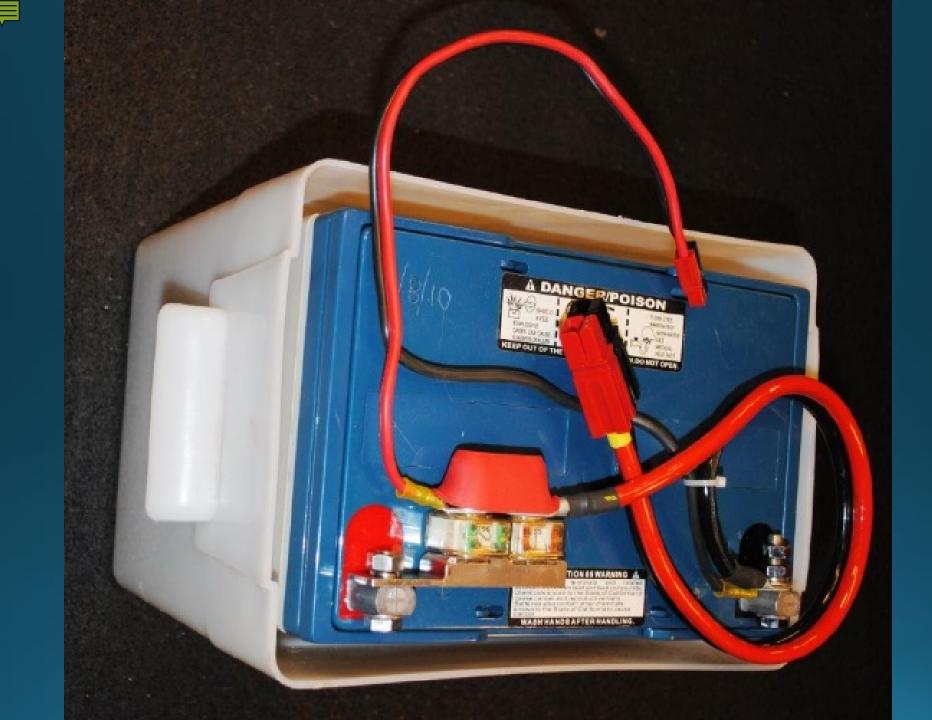
Voltage drop

- If you are losing a half of a volt at the radio due to resistive loss in the battery cable, not only are you losing output power, but you are also losing run time.
- In house wiring, the maximum current for a given wire gauge is based on heat buildup due to the wire's resistance. Most hams know that 12 AWG house wire is good for 20 A.
- But when the wire is used to power a 12V DC radio, the voltage drop of a 12 AWG wire may preclude its use, and you probably need a larger wire size if you are going more than a few feet.



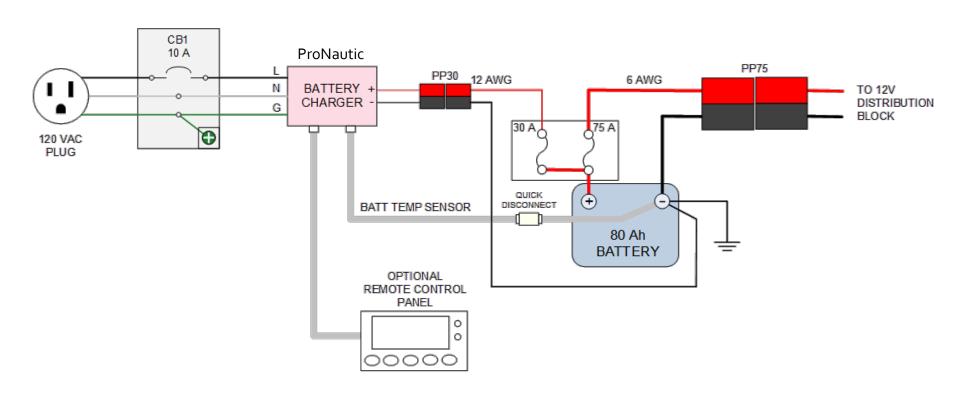








Block diagram of power supply for 100-W HF station



Lithium batteries

All the same hazards as lead-acid batteries, plus:

- Thermal runaway in li-ion batteries. No li-ion battery should be considered immune to this failure mode.
- Ingredients are very, very nasty stuff. If the battery case is compromised, water ingress may cause an explosion.
- Lithium primary batteries behave badly when current flows into the battery instead of out of it. High pressure may cause the battery to rupture. Reverse current flow can occur if a battery in a string is installed backwards, or if two batteries of unequal charge are connected in parallel.

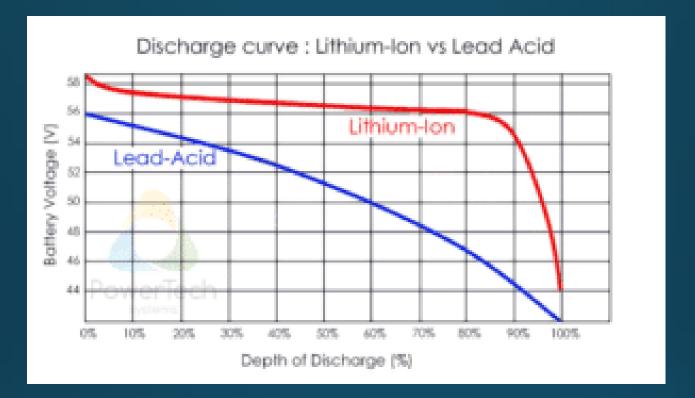
LiFePo Batteries

- Are very popular as lead-acid battery substitutes.
- Are not subject to thermal runaway that plagues Liion batteries.
 - ---but---
- Vulnerable to same hazardous situations as leadacid batteries – with a vengeance
- Extremely sensitive to reverse current and overcharging. Always use the recommended charger.

LiFePo Batteries

- Never parallel LiFePo batteries to get increased capacity.
 Most manufacturers recommend not to place LiFePo
 batteries in series, either, because they accept charge
 unequally, and the first battery in the string will be
 overcharged while the charger continues to deliver
 charging current to bring the rest of the string up to
 terminal voltage.
- LiFePo batteries have a higher terminal voltage (14.6 vs. 13.5 V) when fully charged than the corresponding lead acid battery, and they hold their terminal voltage through most of the discharge cycle. The voltage drops off sharply when the battery approaches the end of its useful capacity. That can be useful in some applications, but it may place additional stress on your equipment.





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Drugstore alkaline batteries

- Even a alkaline battery stores enough energy to cause a second-degree burn is short-circuited.
- The design of the two exposed terminals of a 9V battery in proximity to one another can easily be shortcircuited by coming into contact with any conductive material
- Materials are not particularly harmful. Spent batteries should be placed in regular trash. But put tape over the terminals to be safe. Trash fires have been started by alkaline batteries.



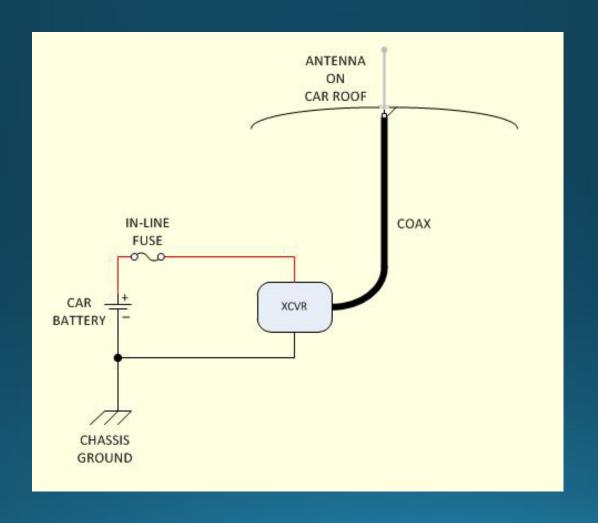




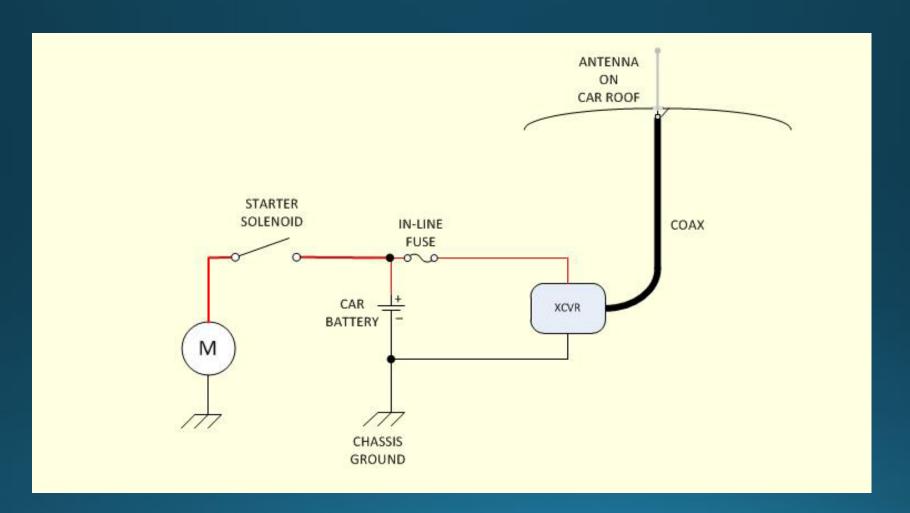


- Car batteries hold lots of energy shorting a battery could cause a fire.
- Special requirements for safe car wiring:
 - Fuse both positive and negative leads.
 - Connect radio's negative lead to negative terminal or engine block ground strap.
 - Use grommets or protective sleeves to protect wires.
 - Don't assume all metal in the car is grounded; modern cars are as much plastic as metal.

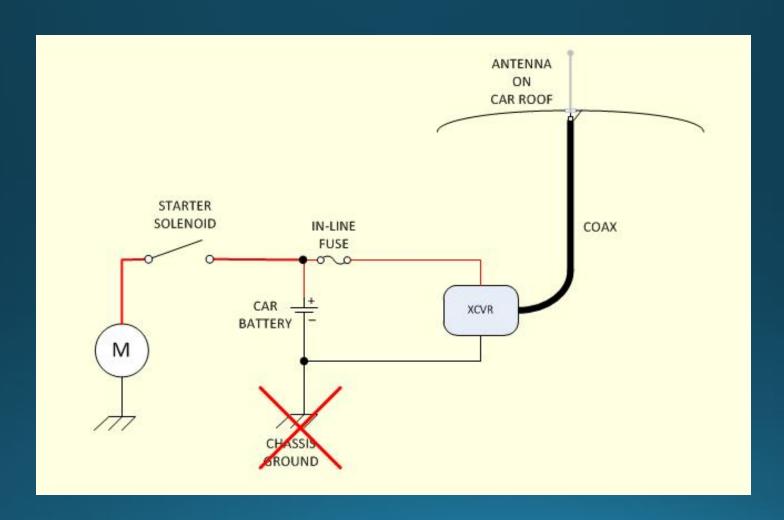




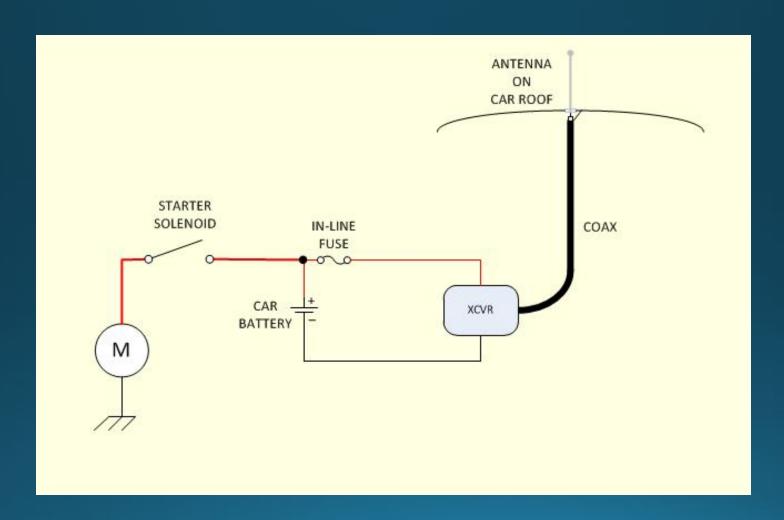




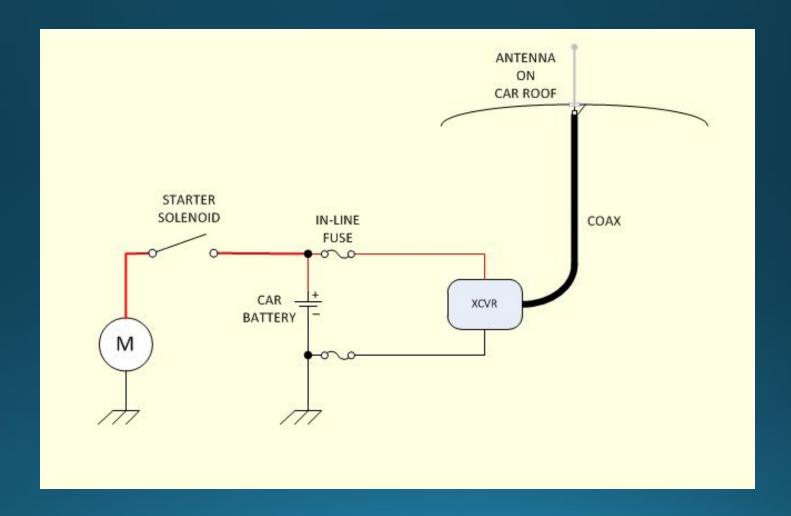












This is not only an issue in mobile radio installations!

- Any time you have high- and low-current devices sharing a common power supply, this problem can occur if the two devices share a signal ground in common.
- Consider a HF rig with any of the following accessories:
 - External keyer
 - External automatic antenna tuner
 - Signalink or similar sound card interface
- If the high-powered radio loses its 12V power supply negative connection, that 20A transmit current is going to flow back to the battery via the accessory.

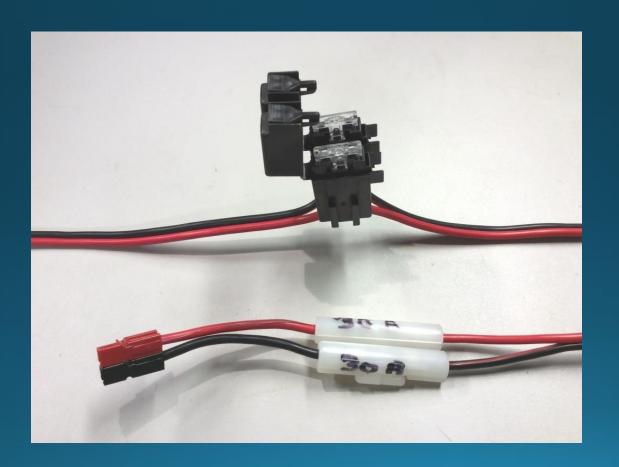
12V outlet strips – two design weaknesses to be concerned with



- Can plug a low-current device into a high-current fuse by accident
- Negative leads of low-current devices are not fused!

A better idea ...

Inline fuses close to power source connector





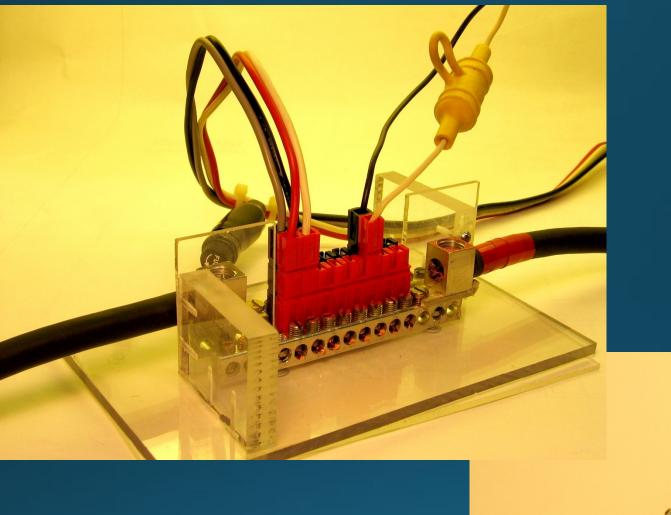
Powerpole distribution blocks



DIY distribution block # 1







DIY distribution block # 2

Jumping batteries safely

- Connect the jumper cables to the weak battery first (keeping the clamps on the other end from shorting together). Always connect the red cable to the positive terminal.
- Then connect the other end of the red jumper cable to the fresh battery – double-check the polarity
- Finally, connect the black (negative) cable to the engine block or vehicle chassis as far away from the battery as possible (in case there is hydrogen gas built up near the battery, which could be ignited by the spark).
- Disconnect the black cable first after the vehicle starts.

Self-powered jump-starter

- Jumper cable w/ reverse polarity protection
- 5V USB outlets
- Laptop power supply –
 12, 16 or 19V
- Flashlight / strobe
- Wall charger and cig lighter charging cable
- Various models with lithium-ion batteries between 12 and 22 Ah capacity



Self-powered jump-starter

- Make your own Powerpole adapter
- Most models require a female EC5 battery connector to plug into the main output
- Two-pack of connectors on Amazon for \$11
- Check to see which size fits your unit (connector size is embossed in connector housing)
- Put inline fuse in positive lead



Female connector has straight barrel contacts

Questions/discussion